

HW day 3

H. not X

5. Calculate the following using the correct number of significant figures.

a) $2.34 \times 10^8 + 9.22 \times 10^7$ 3.26×10^8 b) $9.1 \times 10^3 - 1.6 \times 10^3$ $7500 = 7.5 \times 10^3$

6. Calculate the following using the correct number of significant figures.

a) $(1.54 \times 10^5)(3.5 \times 10^6)$ $5.39 \times 10^{11} = 5.4 \times 10^{11}$

b) $(7.9 \times 10^{34}) \div (8.32 \times 10^{23})$ $9.495 \times 10^{10} = 9.5 \times 10^{10}$

7. Express the following numbers in scientific notation.

a) 810,000 g 8.1×10^5 g b) 0.000634 g 6.34×10^{-4} g c) 60,000,000 g 6×10^7 g

8. State the number of significant digits in the following measurements.

a) 3218 kg 4 b) 60.080 kg 5 c) 0.000534 kg 3

9. Add/Subtract as indicated and round the answer using the correct number of significant digits.

a) $43.218 \text{ g} + 2.7 \text{ g}$ $45.918 \text{ g} = 45.9 \text{ g}$ d) $27.34 \text{ g} + 6.90 \text{ g}$ $34.24 \text{ g} = 34.24 \text{ g}$

b) $1.07 \text{ km} + 0.608 \text{ km}$ $1.678 \text{ km} = 1.68 \text{ km}$ e) $14.325 \text{ m} - 8.92 \text{ m}$ $5.405 \text{ m} = 5.41 \text{ m}$

c) $186.4 \text{ kg} - 57.83 \text{ kg}$ $128.57 \text{ kg} = 128.6 \text{ kg}$ f) $85.26 \text{ g} + 4.7 \text{ g}$ $89.96 \text{ g} = 90.0 \text{ g}$

10. Multiply/Divide as indicated and round the answer using the correct number of significant digits.

a) $(5.108 \text{ m})(4.2107 \text{ m})$ 21.50823 m^2 21.51 m^2

d) $74.50 \div 4.99$ 14.9299 14.9

b) $(1.67 \times 10^{-2} \text{ m})(8.5 \times 10^{-6} \text{ m})$ $1.4 \times 10^{-7} \text{ m}^2$

e) $0.32 \times 14.50 \times 120$ 51.60 or 5.16×10^2

c) $(2.6 \times 10^4 \text{ cm})(9.4 \times 10^6 \text{ cm})$ $2.444 \times 10^{11} \text{ cm}^2$ $2.4 \times 10^{11} \text{ cm}^2$

f) $24.1 \div 0.005$ 4820 5000